JEE MAIN ONLINE PAPER 2021

Held on July 27, 2021 (Evening)

Instructions

1. This test will be a 3 hours Test.
2. This test consists of Physics, Chemistry and Mathematics questions with equal weightage of 100 marks.
3. Each question is of 4 marks.
4. In the question paper consisting of Physics (Q.no. 1 to 30), Chemistry (Q.no. 31 to 60) and Mathematics (Q.no. 61 to 90). There are two sections for each subject (Section-A : MCQ Type & Section-B : Numerical Response Type). Section-A consists of 20 multiple choice questions & Section-B consists of 10 Numerical Value type Questions. **Candidates have a choice to Answer 5 out of the 10 numerical value answer based questions per section.**
5. There will be only one correct choice in the given four choices in Section-A. For each question 4 marks will be awarded for correct choice, 1 mark will be deducted for incorrect choice and zero mark will be awarded for not attempted question. For Section-B questions 4 marks will be awarded for correct answer and zero for unattempted and incorrect answer.
6. Any textual, printed or written material, mobile phones, calculator etc. is not allowed for the students appearing for the test.
7. All calculations/written work should be done in the rough sheet provided.

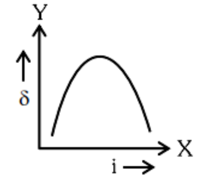
PHYSICS

Section -A

Q.1 An electron and proton are separated by a large distance. The electron starts approaching the proton with energy 3 eV. The proton captures the electrons and forms a hydrogen atom in second excited state. The resulting photon is incident on a photosensitive metal of threshold wavelength 4000 Å. What is the maximum kinetic energy of the emitted photoelectron?

(1) 7.61 eV
(2) 1.41 eV
(3) 3.3 eV
(4) No photoelectron would be emitted

Q.2 The expected graphical representation of the variation of angle of deviation 'δ' with angle of incidence 'i' in a prism is :

(1)


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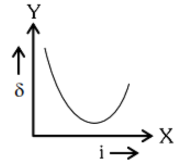

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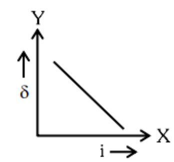

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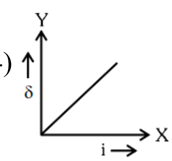


Q.3 A raindrop with radius R = 0.2 mm falls from a cloud at a height h = 2000 m above the ground. Assume that the drop is spherical throughout its fall and the force of buoyancy may be neglected, then the terminal speed attained by the raindrop is :

[Density of water $f_w = 1000$ kg m^{-3} and Density of air $f_a = 1.2$ kg m^{-3}, g = $10 m/s^2$
Coefficient of viscosity of air = 1.8×10^{-5} Nsm^{-2}]

(1) 250.6 ms^{-1}
(2) 43.56 ms^{-1}
(3) 4.94 ms^{-1}
(4) 14.4 ms^{-1}

Q.4 One mole of an ideal gas is taken through an adiabatic process where the temperature rises from 27°C to 37°C. If the ideal gas is composed of polyatomic molecule that has 4 vibrational modes, which of the following is true?

[R = 8.314 J mol^{-1} k^{-1}]

(1) work done by the gas is close to 332 J
(2) work done on the gas is close to 582 J
(3) work done by the gas is close to 582 J
(4) work done on the gas is close to 332 J

Q.5 An object of mass 0.5 kg is executing simple harmonic motion. It amplitude is 5 cm and time period (T) is 0.2 s. What will be the potential energy of the object at an instant $t = \frac{T}{4}$ s starting from mean position, Assume that the initial phase of the oscillation is zero.

(1) 0.62 J (2) 6.2×10^{-3} J
(3) 1.2×10^{3} J (4) 6.2×10^{3} J

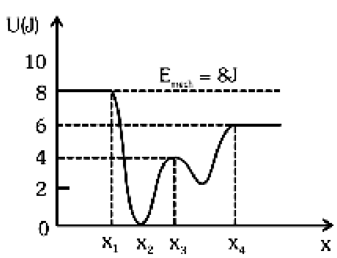
Q.6 Match **List I** with **List II.**

List-I	List-II
(a) Capacitance, C	(i) $M^1L^1T^{-3}A^{-1}$
(b) Permittivity of free space, ε_0	(ii) $M^{-1}L^{-3}T^4A^2$
(c) Permeability of free space, μ_0	(iii) $M^{-1}L^{-2}T^4A^2$
(d) Electric field, E	(iv) $M^1L^1T^{-2}A^{-2}$

Choose the correct answer from the options given below

(1) (a) → (iii), (b) → (ii), (c) → (iv), (d) → (i)
(2) (a) → (iii), (b) → (iv), (c) → (ii), (d) → (i)
(3) (a) → (iv), (b) → (ii), (c) → (iii), (d) → (i)
(4) (a) → (iv), (b) → (iii), (c) → (ii), (d) → (i)

Q.7 Given below is the plot of a potential energy function U(x) for a system, in which a particle is in one dimensional motion, while a conservative force F(x) acts on it. Suppose that E_{mech} = 8 J, the incorrect statement for this system is :



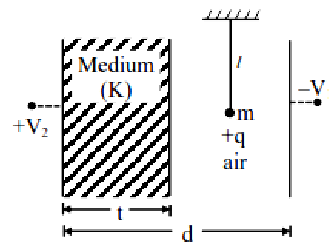
[where K.E. = kinetic energy]

(1) at $x > x_4$, K.E. is constant throughout the region.
(2) at $x < x_1$, K.E. is smallest and the particle is moving at the slowest speed.
(3) at $x = x_2$, K.E. is greatest and the particle is moving at the fastest speed.
(4) at $x = x_3$, K.E. = 4 J.

Q.8 A 100 Ω resistance, a 0.1 μF capacitor and an inductor are connected in series across a 250 V supply at variable frequency. Calculate the value of inductance of inductor at which resonance will occur. Given that the resonant frequency is 60 Hz.

(1) 0.70 H (2) 70.3 mH
(3) 7.03×10^{-5} H (4) 70.3 H

Q.9 A simple pendulum of mass 'm', length 'l' and charge '+q' suspended in the electric field produced by two conducting parallel plates as shown. The value of deflection of pendulum in equilibrium position will be



(1) $\tan^{-1}\left[\frac{q}{mg}\times\frac{C_1(V_2-V_1)}{(C_1+C_2)(d-t)}\right]$

(2) $\tan^{-1}\left[\frac{q}{mg}\times\frac{C_2(V_2-V_1)}{(C_1+C_2)(d-t)}\right]$

(3) $\tan^{-1}\left[\frac{q}{mg}\times\frac{C_2(V_1+V_2)}{(C_1+C_2)(d-t)}\right]$

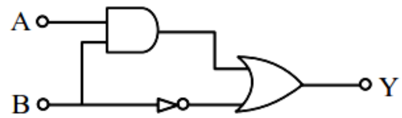
(4) $\tan^{-1}\left[\frac{q}{mg}\times\frac{C_1(V_1+V_2)}{(C_1+C_2)(d-t)}\right]$

Q.10 Two Carnot engines A and B operate in series such that engine A absorbs heat at T_1 and rejects heat to a sink at temperature T. Engine B absorbs half of the heat rejected by Engine A and rejects heat to the sink at T_3. When work done in both the cases is equal, to value of T is

(1) $\frac{2}{3}T_1+\frac{3}{2}T_3$ (2) $\frac{1}{3}T_1+\frac{2}{3}T_3$

(3) $\frac{3}{2}T_1+\frac{1}{3}T_3$ (4) $\frac{2}{3}T_1+\frac{1}{3}T_3$

Q.11 Find the truth table for the function Y of A and B represented in the following figure



(1)

A	B	Y
0	0	0
0	1	1
1	0	0
1	1	0

(2)

A	B	Y
0	0	1
0	1	0
1	0	1
1	1	1

(3)

A	B	Y
0	0	0
0	1	0
1	0	0
1	1	1

(4)

A	B	Y
0	0	0
0	1	1
1	0	1
1	1	1

Q.12 Figure A and B shown two long straight wires of circular cross-section (a and b with a < b), carrying current I which is uniformly distributed across the cross-section. The magnitude of magnetic field B varies with radius r and can be represented as :

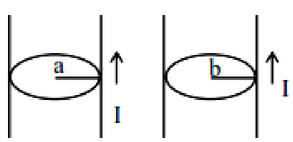
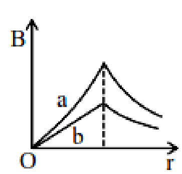
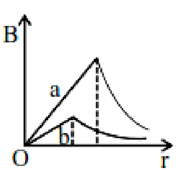
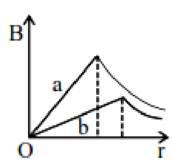
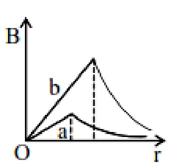


Fig. A Fig. B

(1) 

(2) 

(3) 

(4) 

Q.13 Two identical particles of mass 1 kg each go round a circle of radius R, under the action of their mutual gravitational attraction. The angular speed of each particle is :

(1) $\sqrt{\frac{G}{2R^3}}$ (2) $\frac{1}{2}\sqrt{\frac{G}{R^3}}$

(3) $\frac{1}{2R}\sqrt{\frac{1}{G}}$ (4) $\sqrt{\frac{2G}{R^3}}$

Q.14 Consider the following statements :

A. Atoms of each element emit characteristics spectrum.

B. According to Bohr's Postulate, an electron in a hydrogen atom, revolves in a certain stationary orbit.

C. The density of nuclear matter depends on the size of the nucleus.

D. A free neutron is stable but a free proton decay is possible.

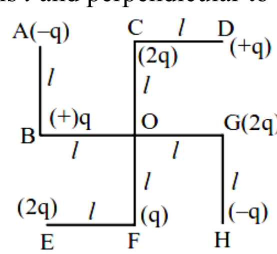
E. Radioactivity is an indication of the instability of nuclei.

Choose the correct answer from the options given below :

(1) A, B, C, D and E (2) A, B and E only

(3) B and D only (4) A, C and E only

Q.15 What will be the magnitude of electric field at point O as shown in figure? Each side of the figure is l and perpendicular to each other?



(1) $\frac{1}{4\pi\varepsilon_0}\frac{q}{\ell^2}$ (2) $\frac{1}{4\pi\varepsilon_0}\frac{q}{(2\ell^2)}(2\sqrt{2}-1)$

(3) $\frac{q}{4\pi\varepsilon_0(2\ell)^2}$ (4) $\frac{1}{4\pi\varepsilon_0}\frac{2q}{2\ell^2}(\sqrt{2})$

Q.16 A physical quantity 'y' is represented by the formula $y = m^2 r^{-4} g^x l^{-\frac{3}{2}}$. If the percentage errors found in y, m, r, l and g are 18, 1, 0. 5, 4 and p respectively, then find the value of x and p.

(1) 5 and ± 2 (2) 4 and ± 3

(3) $\frac{16}{3}$ and $\pm\frac{3}{2}$ (4) 8 and ± 2

Q.17 An automobile of mass 'm' accelerates starting from origin and initially at rest, while the engine supplies constant power P. The position is given as a function of time by :

(1) $\left(\frac{9P}{8m}\right)^{\frac{1}{2}} t^{\frac{3}{2}}$ (2) $\left(\frac{8P}{9m}\right)^{\frac{1}{2}} t^{\frac{3}{2}}$

(3) $\left(\frac{9P}{8m}\right)^{\frac{1}{2}} t^{\frac{3}{2}}$ (4) $\left(\frac{8P}{9m}\right)^{\frac{1}{2}} t^{\frac{3}{2}}$

Q.18 The planet Mars has two moons, if one of them has a period 7 hours, 30 minutes and an orbital radius of 9.0×10^3 km. Find the mass of Mars.

$$\left\{\text{Given } \frac{4\pi^2}{G} = 6\times10^{11}\,N^{-1}m^{-2}kg^2\right\}$$

(1) 5.96×10^{19} kg (2) 3.25×10^{21} kg
(3) 7.20×10^{25} kg (4) 6.00×10^{23} kg

Q.19 A particle of mass M originally at rest is subjected to a force whose direction is constant but magnitude varies with time according to the relation

$$F = F_0\left[1-\left(\frac{t-T}{T}\right)^2\right]$$

Where F_0 and T are constants. The force acts only for the time interval 2T. The velocity v of the particle after time 2T is :

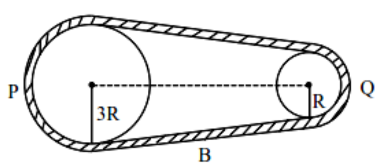
(1) $2F_0T/M$ (2) $F_0T/2M$
(3) $4F_0T/3M$ (4) $F_0T/3/M$

Q.20 The resistance of a conductor at 15°C is 16 Ω and at 100°C is 20Ω. What will be the temperature coefficient of resistance of the conductor ?

(1) $0.010°C^{-1}$ (2) $0.033°C^{-1}$
(3) $0.003°C^{-1}$ (4) $0.042°C^{-1}$

Section -B

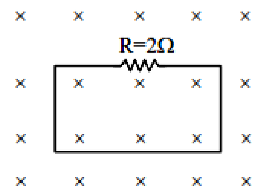
Q.21 In the given figure, two wheels P and Q are connected by a belt B. The radius of P is three times as that of Q. In case of same rotational kinetic energy, the ratio of rotational inertias $\left(\frac{I_1}{I_2}\right)$ will be x : 1. The value of x will be _____.



Q.22 The difference in the number of waves when yellow light propagates through air and vacuum columns of the same thickness is one. The thickness of the air column is _______ mm. [Refractive index of air = 1.0003, wavelength of yellow light in vacuum = 6000 Å]

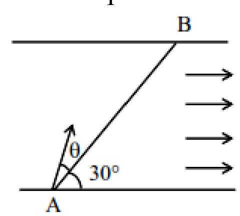
Q.23 The maximum amplitude for an amplitude modulated wave is found to be 12V while the minimum amplitude is found to be 3V. The modulation index is 0.6x where x is ______.

Q.24 In the given figure the magnetic flux through the loop increases according to the relation $\phi_B(t) = 10t^2 + 10t$, where ϕ_B is in milliwebers and t is in seconds. The magnitude of current through R = 2Ω resistor at t = 5 s is _______ m A.



Q.25 A particle executes simple harmonic motion represented by displacement function as $x(t) = A \sin(\omega t + \phi)$ If the position and velocity of the particle at t = 0 s are 2 cm and 2ω cm s^{-1} respectivley, then its amplitude is $x\sqrt{2}$ cm where the value of x is ______.

Q.26 A swimmer wants to cross a river from point A to point B. Line AB makes an angle of 30° with the flow of river. Magnitude of velocity of the swimmer is same as that of the river. The angle θ with the line AB should be ____°, so that the swimmer reaches point B.



Q.27 For the circuit shown, the value of current at time t = 3.2 s will be ______ A.

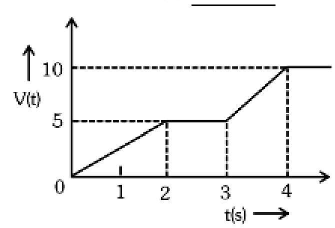


Figure 1

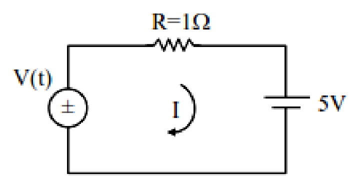
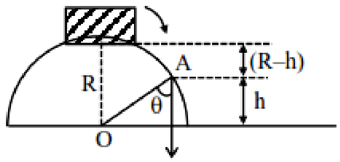


Figure-2

[Voltage distribution V(t) is shown by Fig. (1) and the circuit is shown in Fig. (2)]

Q.28 A small block slides down from the top of hemisphere of radius R = 3 m as shown in the figure. The height 'h' at which the block will lose contact with the surface of the sphere is ____m. (Assume there is no friction between the block and the hemisphere)



Q.29 The K_α X-ray of molybdenum has wavelength 0.071 nm. If the energy of a molybdenum atoms with a K electron knocked out is 27.5 keV, the energy of this atom when an L electron is knocked out will be _______ keV. (Round off to the nearest integer)
[$h = 4.14 \times 10^{-15}$ eVs, $c = 3 \times 10^{8}$ ms^{-1}]

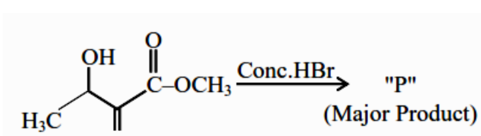
Q.30 The water is filled upto height of 12 m in a tank having vertical sidewalls. A hole is made in one of the walls at a depth 'h' below the water level. The value of 'h' for which the emerging stream of water strikes the ground at the maximum range is ___ m.

CHEMISTRY

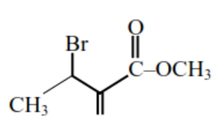
Section -A

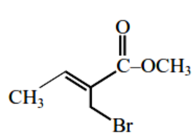
Q.31 Which one of the following set of elements can be detected using sodium fusion extract ?
(1) Sulfur, Nitrogen, Phosphorous, Halogens
(2) Phosphorous, Oxygen, Nitrogen, Halogens
(3) Nitrogen, Phosphorous, Carbon, Sulfur
(4) Halogens, Nitrogen, Oxygen, Sulfur

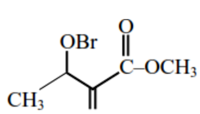
Q.32

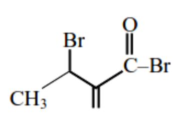


Consider the above reaction, the major product "P" formed is :-

(1) 

(2) 

(3) 

(4) 

Q.33 The number of neutrons and electrons, respectively, present in the radioactive isotope of hydrogen is :-
(1) 1 and 1 (2) 3 and 1
(3) 2 and 1 (4) 2 and 2

Q.34 Match List- I with List II.

List - I	List - II
(a) Li	(i) photoelectric cell
(b) Na	(ii) absorbent of CO_2
(c) K	(iii) coolant in fast breeder nuclear reactor
(d) Cs	(iv) treatment of cancer
	(v) bearings for motor engines

Choose the **correct** answer from the options given below :
(1) (a) - (v), (b) - (i), (c) - (ii), (d) - (iv)
(2) (a) - (v), (b) - (ii), (c) - (iv), (d) - (i)
(3) (a) - (iv), (b) - (iii), (c) - (i), (d) - (ii)
(4) (a) - (v), (b) - (iii), (c) - (ii), (d) - (i)

Q.35 Given below are two statement : one is labelled as **Assertion A** and the other is labelled as **Reason R.**
Assertion A : $SO_2(g)$ is adsorbed to a large extent than $H_2(g)$ on activated charcoal.
Reason R : $SO_2(g)$ has a higher critical temperature than $H_2(g)$.
In the light of the above statements, choose the most appropriate answer from the options given below.
(1) Both **A** and **R** are correct but **R** is not the correct explanation of **A**
(2) Both **A** and **R** are correct and **R** is the correct explanation of **A.**
(3) **A** is not correct but **R** is correct.
(4) **A** is correct but **R** is not correct.

Q.36 The **CORRECT** order of first ionization enthalpy is :
(1) Mg < S < A1 < P (2) Mg < A1 < S < P
(3) A1 < Mg < S < P (4) Mg < A1 < P < S

Q.37 Given below are two statement :
Statements I : Hyperconjugation is a permanent effect.
Statements II : Hyperconjugation in ethyl cation $(CH_3-\overset{+}{C}H_2)$ involves the overlapping of $C_{sp^2}-H_{1s}$ bond with empty 2p orbital of other carbon.

Choose the correct option :

(1) Both **Statement I** and **Statement II** are false

(2) **Statement I** is incorrect but **Statement II** is true

(3) **Statement I** is correct but **Statement II** are false.

(4) Both **Statement I** and **Statement II** are true.

Q.38 Given below are two **Statements**

Statement I : $[Mn(CN)_6]^{3-}$, $[Fe(CN)_6]^{3-}$ and $[CO(C_2O_4)_3]^{3-}$ are d^2sp^3 hybridesed.

Statement II : $[MnC1_6]^{3-}$, and $[FeF_6]^{3-}$ are paramagnetic and have 4 and 5 unpaired electrons, respectively.

In the light of the above statements, choose the correct answer from the options given below :

(1) **Statement I** is correct but **statement II** is false

(2) Both **statement** I and **statement II** are false

(3) **Statement I** is incorrect but **statement II** is true

(4) Both **statement I** and **statement II** are true

Q.39 To an aqueous solution containing ions such as Al^{3+}, Zn^{2+}, Ca^{2+}, Fe^{3+}, Ni^{2+}, Ba^{2+}, and Cu^{2+} was added conc. HCl, followed by H_2S.

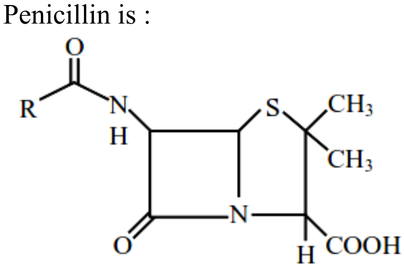
The total number of cations precipitated during this reaction is/are :

(1) 1 (2) 3 (3) 4 (4) 2

Q.40 Given below are two **statements :**

Statements I : Penicillin is a bacteriostatic type antibiotic.

Statement II : The general structure of Penicillin is :



Choose the correct option :

(1) Both **statement I** and **statement II** are false

(2) **Statement I** is incorrect but **statement II** is true

(3) Both **statement I** and **statement II** are true

(4) **Statement I** is correct but **statement II** is false

Q.41 Compound A gives D-Galactose and D-Glucose on hydroglysis. The compound A is :

(1) Amylose (2) Sucrose

(3) Maltose (4) Lactose

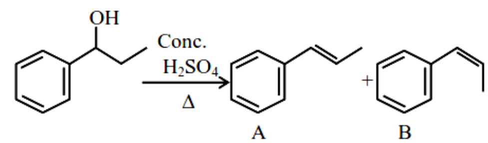
Q.42 $R-CN \xrightarrow[\text{(ii)}H_2O]{\text{(i)DIBAL-H}} R-Y$

Consider the above reaction and identify "Y"

(1) $-CH_2NH_2$ (2) $-CONH_2$

(3) $-CHO$ (4) $-COOH$

Q.43



consider the above reaction, and choose the correct statement :

(1) The reaction is not possible in acidic medium

(2) Both compounds **A** and **B** are formed equally

(3) Compound **A** will be the major product

(4) Compound **B** will be the major product

Q.44 Match List - I with List - II :

List –I (compound)	**List –II (effect/affected species)**
(a) Carbon monoxide	(i) Carcinogenic
(b) Sulphur dioxide	(ii) Metabolized by pyrus plants
(c) Polychlorinated biphenlyls	(iii) Haemoglobin
(d) Oxides of Nitrogen	(iv) Stiffness of flower buds

Choose the correct answer from the options given below :

(1) (a) - (iii), (b) - (iv), (c) - (i), (d) - (ii)

(2) (a) - (iv), (b) - (i), (c) - (iii), (d) - (ii)

(3) (a) - (i), (b) - (ii), (c) - (iii), (d) - (iv)

(4) (a) - (iii), (b) - (iv), (c) - (ii), (d) - (i)

Q.45 If the Thompson model of the atom was correct, then the result of Rutherford's gold foil experiment would have been :

(1) All of the α-particles pass through the gold foil without decrease in speed.

(2) α-Particles are deflected over a wide range of angles.

(3) All αparticles get bounced back by 180°

(4) α-Particles pass through the gold foil deflected by small angles and with reduced speed.

Q.46 Number of Cl = O bonds in chlorous acid, chloric acid and perchloric acid respectively are :

(1) 3, 1 and 1 (2) 4, 1 and 0

(3) 1, 1 and 3 (4) 1, 2 and 3

Q.47 Select the correct statements.

(A) Crystalline solids have long range order.

(B) Crystalline solids are isotropic.

(C) Amorphous solid are sometimes called pseudo solids.

(D) Amorphous solids soften over a range of temperatures.

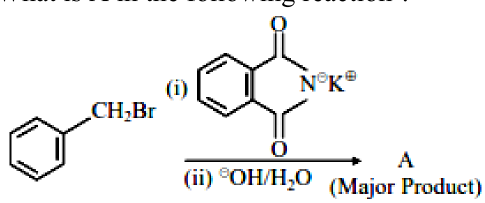
(E) Amorphous solids have a definite heat of fusion.

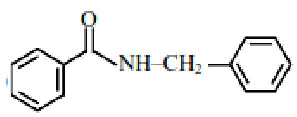
Choose the most appropriate answer from the options given below.

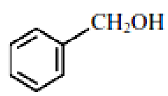
(1) (A), (B), (E) only (2) (B), (D) only

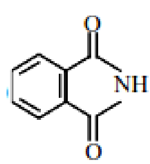
(3) (C), (D) only (4) (A), (C), (D) only

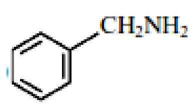
Q.48 What is A in the following reaction ?



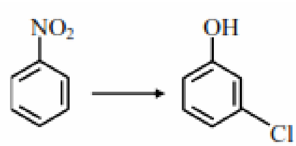
(1) 

(2) 

(3) 

(4) 

Q.49 The correct sequence of correct reagents for the fowling transformation is :-



(1) (i) Fe, HCl (ii) Cl_2, HCl (iii) $NaNO_2$, HCl, 0°C (iv) H_2O/H^+

(2) (i) Fe, HCl (ii) $NaNO_2$, HCl, 0°C (iii) H_2O/H^+ (iv) Cl_2, $FeCl_3$

(3) (i) Cl_2, $FeCl_3$ (ii) Fe, HCl (iii) $NaNO_2$, HCl, 0° C (iv) H_2O/H^+

(4) (i) Cl_2, $FeCl_3$ (ii) $NaNO_2$, HCl, 0° C (iii) Fe, HCl (iv) H_2O/H^+

Q.50 The addition of silica during the extraction of copper from its sulphide ore :-

(1) converts copper sulphide into copper silicate

(2) converts iron oxide into iron silicate

(3) reduces copper sulphide into metallic copper

(4) reduces the melting point of the reaction mixture

Section -B

Q.51 The equilibrium constant for the reaction

$$A(s) \rightleftharpoons M(s) + \frac{1}{2} O_2(g)$$

is $K_p = 4$ At equilibrium, the partial pressure of O_2 is ______ atm. (Round off to the. nearest integer)

Q.52 When 400 mL of 0.2M H_2SO_4 solution is mixed with 600 mL of 0.1 M NaOH solution, the increase in temperature of the final solution is ____ × 10^{-2} K. (Round off to the nearest integer).

[Use : $H^+(aq) + OH^-(aq) \rightarrow H_2O$: $\Delta_\lambda H = -57.1\ kJ\ mol^{-1}$]

Specific heat of $H_2O = 4.18\ J\ K^{-1}\ g^{-1}$

density of $H_2O = 1.0\ g\ cm^{-3}$

Assume no change in volume of solution on mixing.

Q.53 $2SO_2(g) + O_2(g) \rightarrow 2SO_3(g)$

The above reaction is carried out in a vessel starting with partial pressure $P_{SO_2} = 250$m bar, $P_{O_2} = 750$m bar and $P_{SO_3} = 0$ bar. When the reaction is complete, the total pressure in the reaction vessel is ______ m bar. (Round off of the nearest integer).

Q.54 10.0 mL of 0.05 M $KMnO_4$ solution was consumed in a titration with 10.0 mL of given oxalic acid dihydrate solution. The strength of given oxalic acid solution is .... × 10^{-2} g/L. (Round off to the nearest integer)

Q.55 The total number of electrons in all bonding molecular orbitals of O_2^{2-} is …….. .

(Round off to the nearest integer)

Q.56 3 moles of metal complex with formula $Co(en)_2Cl_3$ gives 3 moles of silver chloride on treatment with excess of silver nitrate. The secondary valency of Co in the complex is _____. (Round off to the nearest integer)

Q.57 In a solvent 50% of an acid HA dimerizes and the rest dissociates. The van't Hoff factor of the acid is _____ $\times 10^{-2}$.
(Round off to the nearest integer)

Q.58 The dihedral angle in staggered form of Newman projection of 1, 1, 1-Trichloro ethane is .... degree.
(Round off to the nearest integer)

Q.59 For the first order reaction $A \rightarrow 2B$, 1 mole of reactant A gives 0.2 moles of B after 100 minutes. The half life of the reaction is ...... min. (Round off to the nearest integer).
[Use : ln 2 = 0.69, ln 10 = 2.3
Properties of logarithms : $\ln x^y = y \ln x$;
$\ln\left(\frac{x}{y}\right) = \ln x - \ln y$]
(Round off to the nearest integer)

Q.60 For the cell
$Cu(s)\ |Cu^{2+}(aq)\ (0.1M)\ ||\ Ag^{+}(aq)\ (0.01\ M)\ |\ Ag^{+}(s)$ the cell potential $E_1 = 0.3095$ V
For the cell
$Cu(s)\ |Cu^{2+}(aq)\ (0.01\ M)\ ||\ Ag^{+}.(aq)\ (0.001\ M)\ |\ Ag(s)$
the cell potential = _____ $\times 10^{-2}$ V. (Round off the Nearest Integer).
[Use : $\frac{2.303RT}{F} = 0.059$]

MATHEMATICS

Section -A

Q.61 The point P (a,b) undergoes the following three transformations successively :
(a) reflection about the line y = x.
(b) translation through 2 units along the positive direction of x-axis.
(c) rotation through angle $\frac{\pi}{4}$ about the origin in the anti-clockwise direction.
If the co-ordinates of the final position of the point P are $\left(-\frac{1}{\sqrt{2}}, \frac{7}{\sqrt{2}}\right)$, then the value of 2a + b is equal to :
(1) 13 (2) 9 (3) 5 (4) 7

Q.62 A possible value of 'x', for which the ninth term in the expansion of

$$\left\{3^{\log_3\sqrt{25^{x-1}+7}} + 3^{\left(-\frac{1}{8}\right)\log_3(5^{x-1}+1)}\right\}^{10}$$

in the increasing powers of $3^{\left(-\frac{1}{8}\right)\log_3(5^{x-1}+1)}$ is equal to 180, is :
(1) 0 (2) –1
(3) 2 (4) 1

Q.63 For real number α and $\beta \neq 0$, if the point of intersection of the straight lines

$$\frac{x-\alpha}{1} = \frac{y-1}{2} = \frac{z-1}{3} \text{ and } \frac{x-4}{\beta} = \frac{y-6}{3} = \frac{z-7}{3},$$

lines on the plane x + 2y – z = 8, then $\alpha - \beta$ is equal to :
(1) 5 (2) 9
(3) 3 (4) 7

Q.64 Let $f : \mathbf{R} \rightarrow \mathbf{R}$ be defined as

$f(x + y) + f(x - y) = 2 f(x) f(y)$, $f\left(\frac{1}{2}\right) = -1$.

Then, the value of $\sum_{k=1}^{20} \frac{1}{\sin(k)\sin(k+f(k))}$ is equal to :
(1) $\operatorname{cosec}^2(21)\cos(20)\cos(2)$
(2) $\sec^2(1)\sec(21)\cos(20)$
(3) $\operatorname{cosec}^2(1)\operatorname{cosec}(21)\sin(20)$
(4) $\sec^2(21)\sin(20)\sin(2)$

Q.65 Let C be the set of all complex numbers. Let
$S_1 = \{z \in C : |z - 2| \leq 1\}$ and
$S_2 = \{z \in C : z(1 - i) + \bar{z}(1 - i) \geq 4\}$.
Then, the maximum value of $\left|z - \frac{5}{2}\right|^2$ for $z \in S_1 \cap S_2$ is equal to :
(1) $\frac{3+2\sqrt{2}}{4}$ (2) $\frac{5+2\sqrt{2}}{2}$
(3) $\frac{3+2\sqrt{2}}{2}$ (4) $\frac{5+2\sqrt{2}}{4}$

Q.66 A student appeared in a examination consisting of 8 true–false type question. The student guesses the answers with equal probability. The smallest value of n, so that the probability of guessing at least 'n' correct answers is less than $\frac{1}{2}$, is

(1) 5 (2) 6 (3) 3 (4) 4

Q.67 If $\tan\left(\frac{\pi}{9}\right)$, x, $\tan\left(\frac{7\pi}{18}\right)$ are in arithmetic progression and $\tan\left(\frac{\pi}{9}\right)$, y, $\tan\left(\frac{5\pi}{18}\right)$ are also in arithmetic progression, then $|x - 2y|$ is equal to :

(1) 4 (2) 3 (3) 0 (4) 1

Q.68 Let the mean and variance of the frequency distribution

x :	$x_1 = 2$	$x_2 = 6$	$x_3 = 8$	$x_4 = 9$
f :	4	4	α	β

be 6 and 6.8 respectively. If x_3 is changed from 8 to 7, then the mean for the new will be :

(1) 4 (2) 5 (3) $\frac{17}{3}$ (4) $\frac{16}{3}$

Q.69 The area of the region bounded by $y - x = 2$ and $x^2 = y$ is equal to :-

(1) $\frac{16}{3}$ (2) $\frac{2}{3}$ (3) $\frac{9}{3}$ (4) $\frac{4}{3}$

Q.70 Let $y = y(x)$ be the solution of the differential equation $(x - x^2)dy = (y + yx^2 - 3x^4)dx$, $x > 2$. If $y(3) = 3$, then $y(4)$ is equal to :

(1) 4 (2) 12
(3) 8 (4) 16

Q.71 The value of $\lim_{x\to 0}\left(\frac{x}{\sqrt[8]{1-\sin x} - \sqrt[8]{1+\sin x}}\right)$ is equal to :

(1) 0 (2) 4 (3) – 4 (4) – 1

Q.72 Two sides of a parallelogram are along the lines $4x + 5y = 0$ and $7x + 2y = 0$. If the equation of one of the diagonals of the parallelogram is $11x + 7y = 9$, then other diagonal passes through the point :

(1) (1,2) (2) (2,2)
(3) (2,1) (4) (1,3)

Q.73 Let $\alpha \max_{x\in R}\{8^{2\sin 3x}.4^{4\cos 3x}\}$ and $\beta = \min_{x\in R}\{8^{2\sin 3x}.4^{4\cos 3x}\}$. If $8x^2 + bx + c = 0$ is a quadratic equation whose roots are $\alpha^{1/5}$ and $\beta^{1/5}$ then the value of $c - b$ is equal to :

(1) 42 (2) 47 (3) 43 (4) 50

Q.74 Let $f : [0, \infty) \to [0,3]$ be a function defined by

$$f(x) = \begin{cases} \max\{\sin t : 0 \le t \le x\}, 0 \le x \le \pi \\ 2 + \cos x, \quad x > \pi \end{cases}$$

Then which of the following is true ?

(1) f is continuous everywhere but not differentiable exactly at one point in $(0, \infty)$

(2) f is differentiable everywhere in $(0, \infty)$

(3) f in not continuous exactly at two points in $(0, \infty)$

(4) f is continuous everywhere but not differentiable exactly at two points in $(0, \infty)$

Q.75 Let N be the set of natural number and a relation R on N be defined by

$R = \{(x,y) \in N \times N : x^3 - 3x^2y - xy^2 + 3y^2 = 0\}$

(1) symmetric but neither reflexive nor transitive

(2) reflexive but neither symmetric nor transitive

(3) reflexive and symmetric, but not transitive

(4) an equivalence relation

Q.76 Which of the following is the negation of the statement "for all $M > 0$, then exists $x \in S$ such that $x \ge M$" ?

(1) there exists $M > 0$, such that $x < M$ for all $x \in S$

(2) there exists $M > 0$, there exists $x \in S$ such that $x \ge M$

(3) there exists $M > 0$, there exists $x \in S$ such that $x < M$

(4) there exists $M > 0$, such that $x \ge M$ for all $x \in S$

Q.77 Consider a circle C which touches the y-axis at (0, 6) and cuts off an intercept $6\sqrt{5}$ on the x-axis. Then the radius of the circle C is equal to :

(1) $\sqrt{53}$ (2) 9
(3) 8 (4) $\sqrt{82}$

Q.78 Let $\vec{a}$, $\vec{a},\vec{b}$ $\vec{c}$ be there vectors such that $\vec{a} = \vec{b}\times(\vec{b}\times\vec{c})$. If magnitudes of the vectors $\vec{a},\vec{b}$ and $\vec{c}$ are $\sqrt{2}$,1 and 2 respectively and the angle between $\vec{b}$ and $\vec{c}$ is $\theta\left(0<\theta<\frac{\pi}{2}\right)$, then the value of 1 + tan θ is equal to :

(1) $\sqrt{3}+1$ (2) 2

(3) 1 (4) $\frac{\sqrt{3}+1}{\sqrt{3}}$

Q.79 Let A and B be two 3 × 3 real matrices such that $(A^2 - B^2)$ is invertible matrix. If $A^5 = B^5$ and $A^3B^2 = A^2B^3$, then the value of the determinant of the matrix $A^3 + B^3$ is equal to :

(1) 2 (2) 4 (3) 1 (4) 0

Q.80 Let f : (a,b) → R be twice differentiable function such that $f(x) = \int_a^x g(t)dt$ for a differentiable function g(x). If $f(x) = 0$ has exactly five distinct roots in (a, b), then g(x) = 0 has at least :

(1) twelve roots in (a, b)

(2) five roots in (a, b)

(3) seven roots in (a, b)

(4) three roots in (a, b)

Section -B

Q.81 Let $\vec{a} = \hat{i} - \alpha\hat{j} + \beta\hat{k}$, $\vec{b} = 3\hat{i} + \beta\hat{j} - \alpha\hat{k}$ and $\vec{c} = -\alpha\hat{i} - \hat{j} + \hat{k}$, where α and β are integers.

If $\vec{a}\cdot\vec{b} = -1$ and $\vec{b}\cdot\vec{c} = 10$, then $(\vec{a}\times\vec{b})\cdot\vec{c}$ is equal to ______.

Q.82 The distance of the point P(3, 4, 4) from the point of intersection of the line joining the points. Q(3, –4, –5) and R(2, –3, 1) and the plane 2x + y + z = 7, is equal to______.

Q.83 If the real part of the complex number z $z = \frac{3+2i\cos\theta}{1-3i\cos\theta}$, $\theta \in \left(0,\frac{\pi}{2}\right)$ is zero, then the value of $\sin^2 3\theta + \cos^2\theta$ is equal to _____.

Q.84 Let E be an ellipse whose axes are parallel to the co-ordinates axes, having its center at (3, –4), one focus at (4, – 4) and one vertex at (5, –4). If mx – y = 4, m > 0 is a tangent to the ellipse E, then the value of $5m^2$ is equal to______.

Q.85 If $\int_0^{\pi}(\sin^3 x)e^{-\sin^2 x}dx = \alpha - \frac{\beta}{e}\int_0^1 \sqrt{t}e^t\,dt$, then α + β is equal to ______.

Q.86 The number of real roots of the equation $e^{4x} - e^{3x} - 4e^{2x} - e^x + 1 = 0$ is equal to _____.

Q.87 Let y = y(x) be the solution of the differential equation by = $e^{\alpha x + y}$ dx; α ∈ N. If $y(\log_e 2) = \log_e 2$ and $y(0) = \log_e\left(\frac{1}{2}\right)$, then the value of α is equal to _____.

Q.88 Let n be a non-negative integer. Then the number of divisors of the form "4n + 1" of the number $(10)^{10}.(11)^{11}.(13)^{13}$ is equal to ____.

Q.89 Let $A = \{n \in N \mid n^2 \le n + 10{,}000\}$, B = {3k + 1 | k ∈ N} and C = {2k | k ∈ N}, then the sum of all the elements of the set A ∩ (B – C) is equal to ____.

Q.90 If $A = \begin{vmatrix} 1 & 0 & 1 \\ 0 & 1 & 1 \\ 0 & 0 & 1 \end{vmatrix}$ and $M = A + A^2 + A^3 + \ldots + A^{20}$ then the sum of all the elements of the matrix M is equal to _____.

JEE MAIN ONLINE PAPER 2021

Held on JULY 27, 2021 (Evening)

Hints & Solutions

PHYSICS

Section -A

1. [2] Initially, energy of electron = + 3eV
finally, in 2^{nd} excited state,

energy of electron = $\frac{(13.6eV)}{3^2}$

$= -1.51eV$

Loss in engrgy in emitted as photon,

So, photon energy $\frac{hc}{\lambda} = 4.51eV$

Now, photoelectric effect equation

$KE_{max} = \frac{hc}{\lambda} - \phi = 4.51 - \left(\frac{hc}{\lambda_{th}}\right)$

$= 4.51\ eV - \frac{12400eVÅ}{4000Å}$

$= 1.41\ eV$

2. [2] Standard graph between angle of deviation and incident angle.

3. [3] At terminal speed

$a = 0$

$F_{net} = 0$

$mg = F_v = 6\pi\eta Rv$

$v = \frac{mg}{6\pi\eta Rv}$

$v = \frac{\rho_w \frac{4\pi}{3} R^3 g}{6\pi\eta R}$

$= \frac{2\rho_w R^3 g}{9\eta}$

$= \frac{400}{81}$ m/s

$= 4.94$ m/s

4. [2] Since, each vibrational mode, corresponds to two degrees of freedom, hence, f = 3 (trans.) + 3(rot.) + 8 (vib.) = 14

& $\gamma = 1 + \frac{2}{f}$

$\gamma = 1 + \frac{2}{14} = \frac{8}{7}$

$W = \frac{nR\Delta T}{\gamma - 1} = -582$

As W < 0. Work is done on the gas.

5. [1] $T = 2\pi\sqrt{\frac{m}{k}}$

$0.2 = 2\pi\sqrt{\frac{0.5}{k}}$

$k = 50\pi^2$

≈ 500

$x = A\sin(\omega t + \phi)$

$= 5cm\ \sin\left(\frac{\omega T}{4} + 0\right)$

$= 5cm\ \sin\left(\frac{\pi}{2}\right)$

$= 5cm$

$PE = \frac{1}{2}kx^2$

$= \frac{1}{2}(500)\left(\frac{5}{100}\right)^2$

$= 0.6255$

6. [1] $q = CV$

$[C] = \left[\frac{q}{V}\right] = \frac{(A \times T)^2}{ML^2T^{-2}}$

$= M^{-1}L^{-2}T^4A^2$

$[E] = \left[\frac{F}{q}\right] = \frac{MLT^{-2}}{AT}$

$= MLT^{-3}A^{-1}$

$F = \frac{q_1q_2}{4\pi \epsilon_0 r^2}$

$[\epsilon_0] = M^{-1}L^{-3}T^4A^2$

Speed of light $c = \frac{1}{\sqrt{\mu_0 \epsilon_0}}$

$\mu_0 = \frac{1}{\epsilon_0 c^2}$

$[\mu_0] = \frac{1}{[M^{-1}L^{-3}T^4A^2][LT^{-1}]^2}$

$= [M^1L^1T^{-2}A^{-2}]$

7. [2] $E_{mech.} = 8J$

(A) at $x > x_4$, $U = \text{constant} = 6J$

$K = E_{mech.} - U = 2J = \text{constant}$

(B) at $x < x_1$, $U = \text{constant} = 8J$

$K = E_{mech.} - U = 8 - 8 = 0\ J$

Particle is at rest.

(C) At $x = x_2$, $U = 0 \Rightarrow E_{mech.} = K = 8\ J$

KE is greatest, and particle is moving at fastest speed.

(D) At $x = x_3$, $U = 4\ J$

$U + K = 8\ J$

$K = 4\ J$

8. [4] $C = 0.1\ \mu F = 10^{-7}\ F$

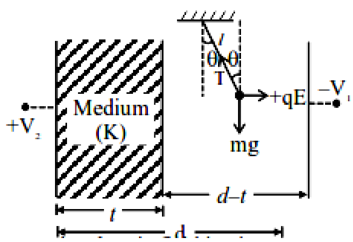
Resonant frequency = 60 Hz

$$\omega_0 = \frac{1}{\sqrt{LC}}$$

$$2\pi f_0 = \frac{1}{\sqrt{LC}} \Rightarrow L = \frac{1}{4\pi^2 f_0^2 C}$$

by putting values $L \simeq 70.3$ Hz.

9. [3]

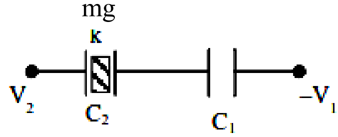


Let E be electric field in air

$T \sin\theta = qE$

$T \cos\theta = mg$

$$\tan\theta = \frac{qE}{mg}$$



$$Q = \left[\frac{C_1C_2}{C_1 + C_2}\right][V_1 + V_2]$$

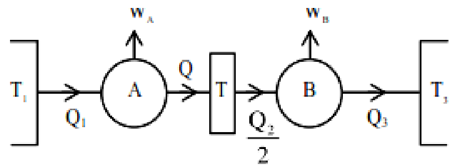
$$E = \frac{Q}{A\,\epsilon_0} = \left[\frac{C_1C_2}{C_1 + C_2}\right]\frac{[V_1 + V_2]}{A\,\epsilon_0}$$

$$C_1 = \frac{\epsilon_0 A}{d - t} \Rightarrow E = \frac{C_2[V_1 + V_2]}{(C_1 + C_2)(d - t)}$$

$$\text{Now } \theta = \tan^{-1}\left[\frac{q.E}{mg}\right]$$

$$\theta = \tan^{-1}\left[\frac{q}{mg} \times \frac{C_2(V_1 + V_2)}{(C_1 + C_2)(d - t)}\right]$$

10. [4]



$$W_A = 1 - \frac{Q_2}{Q_1} = 1 - \frac{T}{T_1} . \Rightarrow \frac{Q_2}{Q_1} = \frac{T}{T_1}$$

$$W_B = 1 - \frac{Q_3}{(Q_2./2)} = 1 - \frac{T_3}{T} \Rightarrow \frac{2Q_3}{Q_2} = \frac{T_3}{T}$$

Now, $W_A = W_B$

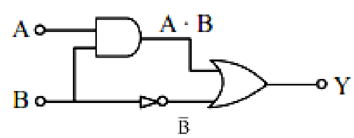
$$Q_1 - Q_2 = \frac{Q_2}{2} - Q_3$$

$$\Rightarrow \frac{2Q_1}{Q_2} + \frac{2Q_3}{Q_3} = 3$$

$$\Rightarrow \frac{2T_1}{T} + \frac{T_3}{T} = 3$$

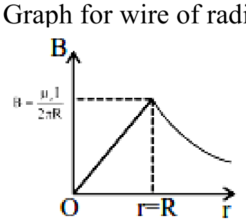
$$\frac{2T_1}{3} + \frac{T_3}{3} = T$$

11. [2]



$Y = A.B + \overline{B}$

12. [3] Graph for wire of radius R :



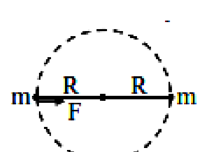
As $b > a$

$B_a > B_b$

$$B_a = \frac{\mu_0 i}{2\pi a}$$

$$B_b = \frac{\mu_0 i}{2\pi b}$$

13. [2]



$$F = \frac{Gm^2}{(2R)^2} = mR\omega^2$$

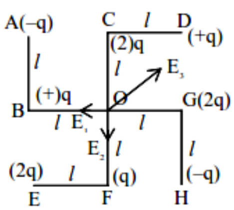
$$\omega = \frac{1}{2}\sqrt{\frac{G}{R^3}}$$

14. [2] (A) True, atom of each element emits characteristic spectrum.
(B) True, according to Bohr's postulates mvr = $\frac{nh}{2\pi}$ and hence electron resides into orbits of specific radius called stationary orbits.
(C) False, density of nucleus is constant
(D) False, A free neutron is unstable decays into proton and electron and antineutrino.
(E) True unstable nucleus show radioactivity.

15. [2] $E_1 = \frac{kq}{\ell^2} = E_2$

$E_3 \frac{kq}{(\sqrt{2}\ell)^2} = \frac{kq}{2\ell^2}$

$E = \frac{\sqrt{2}kq}{\ell^2} - \frac{kq}{2\ell^2} = \frac{kq}{2\ell^2}(2\sqrt{2}-1)$



16. [3] $\frac{\Delta y}{y} = \frac{2\Delta m}{m} + \frac{4\Delta r}{r} + \frac{x\Delta g}{g} + \frac{3}{2}\frac{\Delta \ell}{\ell}$

$18 = 2(1) + 4(0.5) + xp + \frac{3}{2}(4)\ 8 = xp$

By checking from options.

$x = \frac{16}{3}, p = \pm\frac{3}{2}$

17. [4] P = const.

$P = Fv = \frac{mv^2 dv}{dx}$

$\int_0^x \frac{P}{m}dx = \int_0^v v^2 dv$

$\frac{Px}{m} = \frac{v^3}{3}$

$\left(\frac{3Px}{m}\right)^{1/3} = v = \frac{dx}{dt}$

$\left(\frac{3P}{m}\right)^{1/3}\int_0^t dt = \int_0^x x^{-1/3}dx$

$\Rightarrow x = \left(\frac{8P}{9m}\right)^{1/2} t^{3/2}$

18. [4] Option D is correct

$T^2 = \frac{4\pi^2}{GM}.r^3$

$M = \frac{4\pi^2}{G}.\frac{r^2}{T^2}$

by putting values
$M = 6 \times 10^{23}$

19. [3] t = 0, u = 0

$a = \frac{F_0}{M} - \frac{F_0}{MT^2}(t-T)^2 = \frac{dv}{dt}$

$\int_0^v dv = \int_{t=0}^{2T}\left(\frac{F_0}{M} - \frac{F_0}{MT^2}(t-T)^2\right)dt$

$V = \left[\frac{F_0}{M}t\right]_0^{2T} - \frac{F_0}{MT^2}\left[\frac{t^3}{3} - t^2T + T^2t\right]_0^{2T}$

$V = \frac{4F_0T}{3M}$

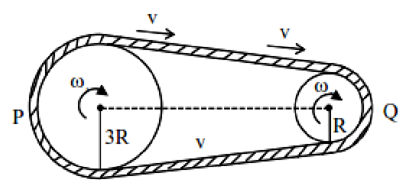
20. [3] $16 = R_0[1 + \alpha(15 - T_0)]$
$20\ R_0[1 + \alpha(100 - T_0)]$
Assuming $T_0 = 0°C$, as a general convention.

$\Rightarrow \frac{16}{20} = \frac{1+\alpha\times 15}{1+\alpha\times 100}$

$\Rightarrow \alpha = 0.003\ °C^{-1}$

Section -B

21. [9]



$\frac{1}{2}I_1(\omega_1)^2 = \frac{1}{2}I_2(\omega_2)^2$

$I_1\left(\frac{v}{3R}\right)^2 = I_2\left(\frac{v}{R}\right)^2$

$\frac{I_1}{I_2} = \frac{9}{1}$

22. [2] Thickness t = nλ
So, $n\lambda_{vac} = (n+1)\lambda_{air}$

$n\lambda = (n+1)\frac{\lambda}{\mu_{air}}$

$n = \frac{1}{\mu_{air}-1} = \frac{10^4}{3}$

t = nλ

$= \frac{10^4}{3} \times 6000$ Å

= 2 mm

23. [1] $A_{max} = A_c + A_m = 12$

$A_{min} = A_c - A_m = 3$

$\Rightarrow A_c = \frac{15}{2}$ & $A_m = \frac{9}{2}$

modulation index $= \frac{A_m}{A_c} = \frac{9/2}{15/2} = 0.6$

$\Rightarrow x = 1$

24. [60] $|\in| = \frac{d\phi}{dt} = 20t + 20$ mV

$|i| = \frac{|\in|}{R} = 10t + 10$ mA

at t = 5

$|i| = 60$ mA

25. [2] $x(t) = A \sin(\omega + \phi)$

$v(t) = A\omega \cos(\omega t + \phi)$

$2 = A \sin \phi$ ......(1)

$2\omega = A\omega \cos \phi$ ......(2)

From (1) and (2)

$\tan \phi = 1$

$\phi = 45°$

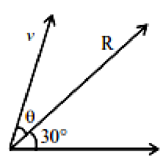
Putting value of ϕ in equation (1)

$2 = A\left\{\frac{1}{\sqrt{2}}\right\}$

$A = 2\sqrt{2}$

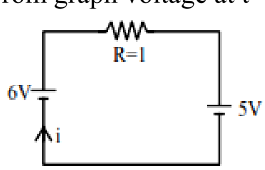
x = 2

26. [30]



Both velocity vectors are of same magnitude therefore resultant would pass exactly midway through them $\theta = 30°$

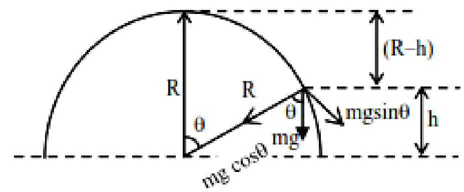
27. [1] From graph voltage at t = 3.2 sec is 6 volt.



$i = \frac{6-5}{1}$

i = 1 A

28. [2]



$mg \cos\theta = \frac{mv^2}{R}$ ....(1)

$\cos\theta = \frac{h}{R}$

Energy conservation

$mg\{R - h\} = \frac{1}{2} mv^2$ ..... (2)

from (1) & (2) $\Rightarrow mg\left\{\frac{h}{R}\right\} = \frac{2mg\{R-h\}}{R}$

$h = \frac{2R}{3} = 2m$

29. [10] $E_{k_\alpha} = E_k - E_L$

$\frac{hc}{\lambda_{k2}} = E_k - E_L$

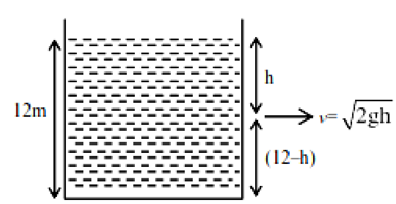
$E_L = E_k - \frac{hc}{\lambda_{k_\alpha}}$

$= 27.5 \text{ KeV} - \frac{12.42\times10^{-7} \text{ eVm}}{0.071\times10^{-9} \text{ m}}$

$E_L = (27.5 - 17.5)$ keV

$= 10$ keV

30. [6]



$R = \sqrt{2gh} \times f\sqrt{\frac{(12-h)\times2}{g}}$

$\sqrt{4h(12-h)} = R$

For maximum R

$\frac{dR}{dh} = 0$

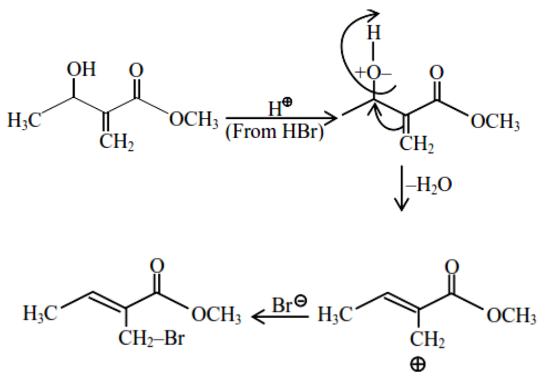
$\Rightarrow h = 6$

CHEMISTRY

Section -A

31. [1] By sodium fusion extract we can detect sulphur, nitrogen, Phosphorous and halogens, because they are converted in to their ionic form with sodium metal.

32. [2]



33. [3] Radioactive isotope of hydrogen is Tritium (^3_1T) No. of neutrons (A –Z) = 3 – 1 = 2
No. of electrons = 1

34. [4] Li makes alloy with Lead to make white metal bearings for motor engines
Liquid Na metal is used as coolant in fast breeder nuclear reactor
K is a very absorbent of CO_2
Cs is used in making photoelectric cell

35. [2] Gases having higher critical temperature absorb to a greater extent.

36. [3] Mg Al P S → IE. order ⇒ Al< Mg < S < P

Valence [Ne] : $\overset{Mg}{3s^2}$ $\overset{A\ell}{3s^2\ 3p^1}$ $\overset{P}{3s^2\ 3p^3}$ $\overset{S}{3s^2\ 3p^4}$

↑ Full Filled Stable (Mg); ↑ Half Filled Stable (P)

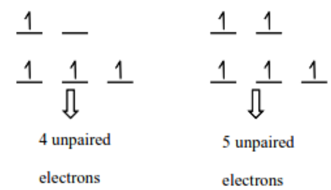
37. [3] **Statement I** It is correct Statement

Statement II : $CH_3 - \overset{\oplus}{C}H_2$ involve $CH_{sp^3} - H_{1s}$ bond with empty 2p orbital hence given statement is false.

38. [4] $[Mn(CN)_6]^{3-}$ $[Fe, (CN)_6]^{3-}$. $[Co(C_2O_4)_3]^{3-}$
⇓
Mn^{3+} CN^- Fe^{3+} , CN^- Co3 + $C_2O_4^{2-}$
d^4 configuration, SFL d^5 configuration, SFL d^6 configuration, Chelating ligand

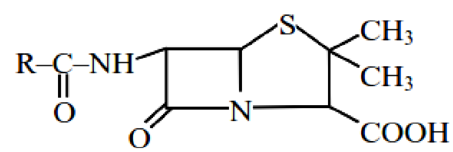
⇒ All will have larger splitting hence d^2sp^3 hybridisation

$[MnCl_6]^{3-}$ and $[FeF_6]^{3-}$
d^4 configuration, $C1^-$ WFL; d^5 configuration, F^- WFL



39. [1] Al^{3+} and Fe^{3+} sulphides dydrolyse in water.
Ni^{2+} and Zn^{2+} require basic medium with H_2S to from ppt
Ca^{2+} and Ba^{2+} sulphides are soluble hence we will receive only CuS ppt.

40. [2] **Statement I :** Pencillin is bactericidal not bacteriostatic hence given statement is false.
Statement II : Structure of pencilline given is correct

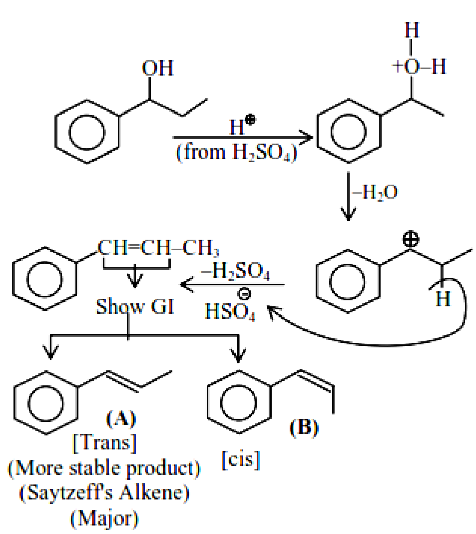


41. [4] **Lactose :** It is a disaccharide of β–D–Galactose and β–D–Glucose with C_1 of galactose and C_4 of glucose link.
Lactose : β–D– Galactose + β–D–Glucose

42. [3] $R{-}C{\equiv}N \xrightarrow[(2)H_2O]{(1)DiBAL{-}H} R{-}\underset{\underset{O}{\|}}{C}{-}H$

Here Y is $-\underset{\underset{O}{\|}}{C}{-}H$ Aldehyde

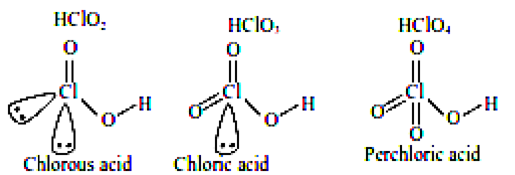
43. [3]



44. [1]

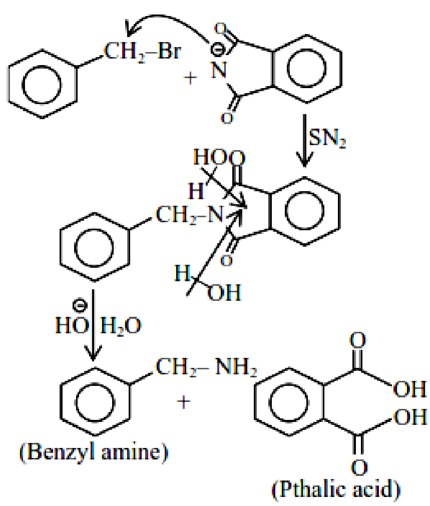
45. [4] As in Thomson model, protons are diffused (charge is not centred) α- particles deviate by small angles and due to repulsion from protons, their speed decreases.

46. [3] Number of Cl = O bonds

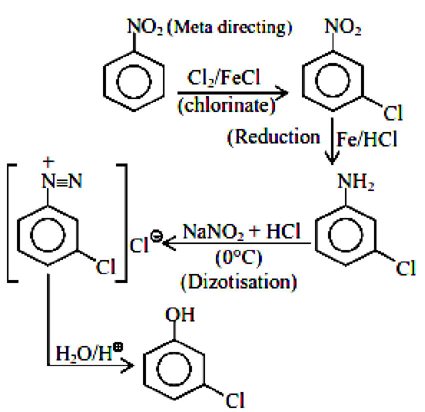


47. [4] (A) Crystalline solids have definite arrangement of constituent particles and have long range order.
(C), (D) Different constituent particles of an amorphous solid have different bond strengths and soften over a range of temperatures.

48. [4]



49. [3]



50. [2] Silica is used to remove FeO impurity from the ore of coper

$FeO + siO_2 \rightarrow FeSiO_3$
iron silicate
(Slag)

Section -B

51. [16] $k_p = Po_2^{1/2} = 4$

$\therefore Po_2 = 16$ bar = 16 atm

52. NTA[2],Allen (82)

$$n_{H^+} = \frac{400 \times 0.2}{1000} \times 2 = 0.16$$

$$n_{OH^-} = \frac{600 \times 0.1}{1000} = 0.06 \text{ (L.R)}$$

Now, heat liberated from reaction
= heat gained by solutions
or, $0.06 \times 57.1 \times 10^3$
$= (1000 \times 1.0) \times 4.18 \times \Delta T$
$\therefore \Delta T = 0.8196$ K
$= 81.96 \times 10^{-2}$ K $\approx 82 \times 10^{-2}$ K

53. [875]

	$2SO_2(g)$	+ $O_2(g)$	$\rightarrow 2SO_3(g)$
initial	250 m bar (L. R.)	750 m bar	O
Final	~~250 m bar~~ 0	~~125 m bar~~ 625 m bar	~~250 m~~ bar 250 m bar

$\therefore$ Final total pressure = 625 + 250 = 875 m bar

54. [1575] $n_{eq}\ KMnO_4 = n_{eq}\ H_2C_2O_4 . 2H_2O$

$$\text{or, } \frac{10 \times 0.05}{1000} \times 5 = \frac{10 \times M}{1000} \times 2$$

$\therefore$ Conc. of oxalic acid solution = 0.125 M
$= 0.125 \times 126$ g/L = 15.75 g/L
$= 1575 \times 10^{-2}$ g/L

55. [10] M. O. Configuration of O_2^{2-} ((18 $\bar{e}$)

$$\sigma 1s^2 \overset{*}{\sigma} 1s^2 \sigma 2s^2 \overset{*}{\sigma} 2s^2 \sigma 2p_z^2 \pi 2p_x^2 = \pi 2p_y^2$$

$$\overset{*}{\pi} 2p_x^2 = \overset{*}{\pi} 2p_y^2$$

Total B. M. O electrons = 10

56. [6] $3[Co(en)_2Cl_2]C\ell + \underset{(excess)}{AgNO_3} \rightarrow \underset{(white\ ppt.)}{3AgCl}$

Secondary valency of Co = 6 (C. N.)

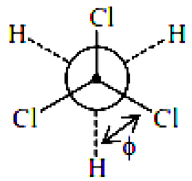
57. [125] $2HA \rightleftharpoons H_2A_2$ $HA \rightleftharpoons H^+ + A$

Initial moles	$a \times \frac{50}{100}$	0	$a \times \frac{50}{100}$	0	0
Final moles	0	0.25 a	0	0.5a	0.5a

$$\text{Now, } i = \frac{\text{final moles}}{\text{initial moles}} = \frac{0.25a + 0.5a + 0.5a}{0.5a + 0.5a}$$

$= 1.25 = 125. \times 10^{-2}$

58. [60] 1,1,1– Trichloro ethane [CCl_3–CH_3]



Dihedral angle(ϕ) = 60°

(Newmonns stqqared form)

59. [690] Official Ans. by NTA[300]

	$A \longrightarrow 2B$	
t = 0	1 mole	0
t = 100 min	1 – x	2x
	= 0.9mol	= 0.2 mol

Now, $t = \dfrac{t_{1/2}}{\ell n2} \times \dfrac{[A_0]}{[A_t]}$

$100 = \dfrac{t_{1/2}}{\ell n2} \times \ell n \dfrac{1}{0.9} \Rightarrow t_{1/2} = 690$ min.

(taking In 3 = 1.11)

Ans. 600 to 700

60. [28] Cell reaction is :

$Cu\ (s) + 2Ag^{+}\ (aq) \rightarrow Cu^{2+}\ (aq) + 2Ag\ (s)$

Now, $E_{cell} = E^{\circ}_{cell} - \dfrac{0.059}{2} \log \dfrac{[Cu^{2+}]}{[Ag^{+}]^2}$ …..(1)

$\therefore E_1 = 0.3095 = E^{\circ}_{cell} - \dfrac{0.059}{2} . \log - \dfrac{0.01}{(0.001)^2}$ ….. (2)

From (1) and (2), $E_2 = 0.28$ V $= 28 \times 10^{-2}$ V

MATHEMATICS

Section -A

61. [2] Image of A(a,b) along y = x is B(b,a).
Translating it 2 units it becomes C(b + 2,a).
Now, applying rotation theorem

$-\dfrac{1}{2} + \dfrac{7}{\sqrt{2}} i = ((b+2) + ai)\left(\cos\dfrac{\pi}{4} + i \sin\dfrac{\pi}{4}\right)$

$\dfrac{-1}{\sqrt{2}} + \dfrac{7}{\sqrt{2}} i = \left(\dfrac{b+2}{\sqrt{2}} - \dfrac{a}{\sqrt{2}}\right) + i\left(\dfrac{b+2}{\sqrt{2}} + \dfrac{a}{\sqrt{2}}\right)$

$\Rightarrow b - a + 2 = -1$ …..(i)

and $b + 2 + a = 7$ …..(ii)

$\Rightarrow a = 4;\ b = 1$

$\Rightarrow 2a + b = 9$

62. [4] $^{10}C_8\ (25^{(x-1)} + 7) \times (5^{(x-1)} + 1)^{-1} = 180$

$\Rightarrow \dfrac{25^{x-1} + 7}{5^{(x-1)} + 1} = 4$

$\Rightarrow \dfrac{t^2 + 7}{t + 1} = 4;$

$\Rightarrow t = 1,\ 3 = 5^{x-1}$

$\Rightarrow x - 1 = 0$ (one of the possible value)

$\Rightarrow x = 1$

63. [4] First line is $(\phi + \alpha, 2\phi + 1, 3\phi + 1)$
and second line is $(q\beta + 4, 3q + 6, 3q + 7)$.

For intersection $\phi + \alpha = q\beta + 4$ … (i)

$2\phi + 1 = 3q + 6$ …(i)

$3\phi + 1 = 3q + 7$ …(iii)

for (ii) & (iii) $\phi = 1, q = -1$

So, from (i) $\alpha + \beta = 3$

Now, point of intersection is $(\alpha + 1,3,4)$

It lies on the plane.

Hence, $\alpha = 5$ & $\beta = -2$

64. [3] $f(x) = \cos \lambda x$

$\because f\left(\dfrac{1}{2}\right) = -1$

So, $-1 = \cos\dfrac{\lambda}{2}$

$\Rightarrow \lambda = 2\pi$

Thus $f(x) = \cos 2\pi x$

Now k is natural number

Thus f(k) = 1

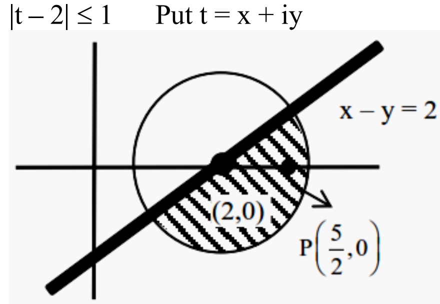
$\sum\limits_{k=1}^{20} \dfrac{1}{\sin k \sin(k+1)} = \dfrac{1}{\sin 1}$

$\sum\limits_{k=1}^{20} \left[\dfrac{\sin(k+1) - k}{\sin k . \sin(k+1)}\right] = \dfrac{1}{\sin 1}$

$\sum\limits_{k=1}^{20} (\cot k - \cot(k+1))$

$= \dfrac{\cot - \cot 21}{\sin 1} = \text{cosec}^2\ 1\ \text{cosec}\ (21).\ \sin 20$

65. [4] $|t - 2| \le 1$ Put $t = x + iy$



$(x - 2)^2 + y^2 \le 1$

Also, $t(\ell + i) + \bar{t}\,(1 - i) \ge 4$

Given $x - y \ge 2$

Let point on circle be $A(2 + \cos\theta, \sin\theta)$

$\theta \in \left[-\dfrac{3\pi}{4}, \dfrac{\pi}{4}\right]$

$(AP)^2 = \left(2 + \cos\theta - \frac{5}{2}\right)^2 + \sin^2\theta$

$= \cos^2\theta - \cos\theta + \frac{1}{4} + \sin^2\theta$

$= \frac{5}{4} - \cos\theta$

For $(AP)^2$ maximum $\theta = -\frac{3\pi}{4}$

$(AP)^2 = \frac{5}{4} + \frac{1}{\sqrt{2}} = \frac{5\sqrt{2}+4}{4\sqrt{2}}$

66. [1] $P(E) < \frac{1}{2}$

$\Rightarrow \sum_{r=n}^{8} {}^8C_r \left(\frac{1}{8}\right)^{8-r}\left(\frac{1}{2}\right)^r < \frac{1}{2}$

$\Rightarrow \sum_{r=n}^{8} {}^8C_r \left(\frac{1}{2}\right)^{8} < \frac{1}{2}$

$\Rightarrow {}^8C_n + {}^8C_{n+1} + \ldots\ {}^8C_8 < 128$

$\Rightarrow 256 - ({}^8C_0 + {}^8C_1 + \ldots + {}^8C_{n-1}) < 128$

$\Rightarrow {}^8C_0 + {}^8C_1 + \ldots + {}^8C_{n-1} > 128$

$\Rightarrow n - 1 \ge 4$

$\Rightarrow n \ge 5$

67. [3] $x = \frac{1}{2}\left(\tan\frac{\pi}{9} + \tan\frac{7\pi}{18}\right)$

and $2y = \tan\frac{\pi}{9} + \tan\frac{5\pi}{18}$

so, $x - 2y = \frac{1}{2}\left(\tan\frac{\pi}{9} + \tan\frac{7\pi}{18}\right)$

$-\left(\tan\frac{\pi}{9} + \tan\frac{5\pi}{18}\right)$

$\Rightarrow |x - 2y| = \left|\frac{\operatorname{con}\frac{\pi}{9} - \tan\frac{\pi}{9}}{2} - \tan\frac{5\pi}{18}\right|$

$= \left|\cot\frac{2\pi}{9} - \cot\frac{2\pi}{9}\right| = 0$

$\left(\text{as} \tan\frac{5\pi}{18} = \cot\frac{2\pi}{9}; \tan\frac{7\pi}{18} = \cot\frac{\pi}{9}\right)$

68. [3] Given $32 + 8\alpha + 9\beta = (8 + \alpha + \beta) \times 6$

$\Rightarrow 2\alpha + 3\beta = 16$ ....(i)

Also, $4 \times 16 + 4 \times \alpha + 9\beta = (8 + \alpha + \beta) \times 6.8$

$\Rightarrow 640 + 40\alpha + 90\beta = 544 + 68\alpha + 68\beta$

$\Rightarrow 28\alpha - 22\beta = 96$

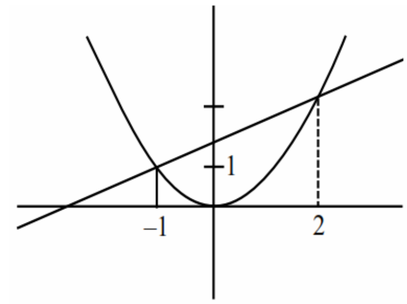
$\Rightarrow 14\alpha - 11\beta = 48$ ....(ii)

from (i) & (ii)

$\alpha = 5$ & $\beta = 2$

so, now mean $= \frac{32 + 35 + 18}{15} = \frac{85}{15} = \frac{17}{3}$

69. [3]



$y - x = 2,\ x^2 = y$

Now, $x^2 = 2 + x$

$\Rightarrow x^2 - x - 2 = 0$

$\Rightarrow (x + 1)(x - 2) = 0$

Area $= \int_{-1}^{2} (2 + x - x^2)$

$= \left|2x + \frac{x^2}{2} - \frac{x^3}{3}\right|_{-1}^{2}$

$= \left(4 + 2 - \frac{8}{3}\right) - \left(-2 + \frac{1}{2} + \frac{1}{3}\right)$

$= 6 - 3 + 2 - \frac{1}{2} = \frac{9}{2}$

70. [2] $(x - x^3)dy = (y + yx^2 - 3x^4)dx$

$\Rightarrow xdy - ydx = (yx^2 - 3x^4)dx + x^3\,dy$

$\Rightarrow \frac{xdy - ydx}{x^2} = (ydx + xdy) - 3x^2dx$

$\Rightarrow d\left(\frac{y}{x}\right) = d(xy) - d(x^3)$

Integrate

$\Rightarrow \frac{y}{x} = xy - x^3 + c$

given $f(3) = 3$

$\Rightarrow \frac{3}{3} = 3 \times 3 - 3^3 + c$

$\Rightarrow c = 19$

$\therefore \frac{y}{x} = xy - x^3 + 19$

at $x = 4$, $\frac{y}{4} = 4y - 64 + 19$

$15y = 4 \times 45$

$\Rightarrow y = 12$

71. [3] $\lim\limits_{x\to 0}\left(\dfrac{x}{\sqrt[8]{1-\sin x}-\sqrt[8]{1+\sin x}}\right)$

$= \lim\limits_{x\to 0}\left(\dfrac{x}{\sqrt[8]{1-\sin x}-\sqrt[8]{1+\sin x}}\right)$

$= \lim\limits_{x\to 0}\left(\dfrac{x}{\sqrt[8]{1-\sin x}-\sqrt[8]{1+\sin x}}\right)$

$\left(\dfrac{\sqrt[8]{1-\sin x}+\sqrt[8]{1+\sin x}}{\sqrt[8]{1-\sin x}+\sqrt[8]{1+\sin x}}\right)$

$\left(\dfrac{\sqrt[4]{1-\sin x}+\sqrt[4]{1+\sin x}}{\sqrt[4]{1-\sin x}+\sqrt[4]{1+\sin x}}\right)$

$\left(\dfrac{\sqrt[2]{1-\sin x}+\sqrt[2]{1+\sin x}}{\sqrt[2]{1-\sin x}+\sqrt[2]{1+\sin x}}\right)$

$\lim\limits_{x\to 0}\left(\dfrac{x}{1-\sin x-(1+\sin x)}\right)$

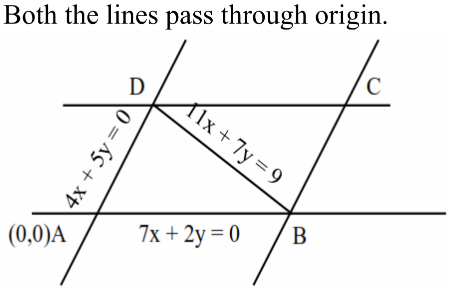
$(\sqrt[8]{1-\sin x}+\sqrt[8]{1+\sin x})\ (\sqrt[4]{1-\sin x}+\sqrt[4]{1+\sin x})$
$(\sqrt[2]{1-\sin x}+\sqrt[2]{1+\sin x})$

$= \lim\limits_{x\to 0}\dfrac{x}{(-2\sin x)}(\sqrt[8]{1-\sin x}+\sqrt[8]{1+\sin x})$

$(\sqrt[4]{1-\sin x}+\sqrt[4]{1+\sin x})\ (\sqrt[2]{1-\sin x}+\sqrt[2]{1+\sin x})$

$= \lim\limits_{x\to 0}\left(-\dfrac{1}{2}\right)(2)(2)(2)\ \left\{\because \lim\limits_{x\to 0}\dfrac{\sin x}{x}=1\right\} = -4$

72. [2] Both the lines pass through origin.



Point D is equal of intersection of $4x + 5y = 0$ & $11x + 7y = 9$

So, coordinates of point D = $\left(\dfrac{5}{3}, -\dfrac{4}{3}\right)$

Also, point B is point of intersection of $7x + 2y = 0$ & $11x + 7y = 9$

So, coordinates of point B = $\left(-\dfrac{2}{3}, \dfrac{7}{3}\right)$

diagonals of parallelogram intersect at middle let middle point of B,D

$\Rightarrow \left(\dfrac{\frac{5}{3}-\frac{2}{3}}{2}, \dfrac{\frac{-4}{3}+\frac{7}{3}}{2}\right) = \left(\dfrac{1}{2}, \dfrac{1}{2}\right)$

equation of diagonal AC

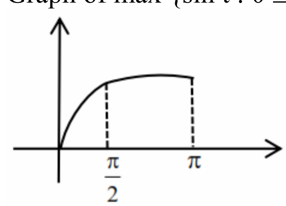
$\Rightarrow (y-0)\ \dfrac{\frac{1}{\alpha}-0}{\frac{1}{\alpha}-0}(\pi - 0)$

$y = x$

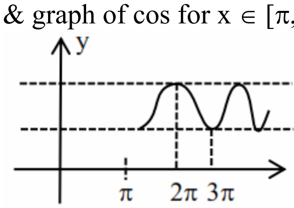
diagonal AC passes through (2, 2).

73. [1] $\alpha = \max\{8^{2\sin 3x} . 4^{4\cos 3x}\}$
$= \max\{2^{6\sin 3x} . 2^{8\cos 3x}\}$
$= \max\{2^{6\sin 3x + 8\cos 3x}\}$
and $\beta = \min\{8^{2\sin 3x} . 4^{4\cos 3x}\} = \min\{2^{6\sin 3x + 8\cos 3x}\}$
Now range of $6 \sin 3x + 8 \cos 3x$
$= \left[-\sqrt{6^2+8^2}, +\sqrt{6^2+8^2}\right] = [-10, 10]$
$\alpha = 2^{10}$ & $\beta = 2^{-10}$
So, $\alpha^{1/5} = 2^2 = 4$
$\Rightarrow \beta^{1/5} = 2^{-2} = 1/4$
quadratic $8x^2 + bx + c = 0$, $c - b = 8 \times$ [(Product of roots)] + (sum of roots)
$= 8\times\left[4\times\dfrac{1}{4}+4+\dfrac{1}{4}\right] = 8\times\left[\dfrac{21}{4}\right] = 42$

74. [2] Graph of $\max\{\sin t : 0 \le t \le x\}$ in $x \in [0, \pi]$

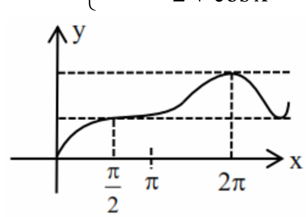


& graph of cos for $x \in [\pi, \infty)$



So graph of

$f(x) = \begin{cases} \max\{\sin t : 0 \le t \le x, & 0 \le x \le \pi \\ 2 + \cos x & x > h \end{cases}$

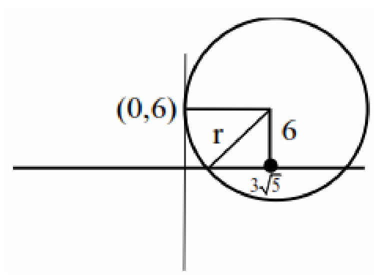


f(x) is differentiable everywhere in $[0, \infty)$

75. [2] $x^2 - 3x^2y - xy^2 + 3y^3 = 0$
$\Rightarrow x(x^2 - y^2) - 3y(x^2 - y^2) = 0$
$\Rightarrow (x - 3y)(x - y)(x + y) = 0$
Now, $x = y\ \forall(x,y) \in N \times N$ so reflexive
But not symmetric & transitive
See, (3,1) satisfies but (1,3) does not. Also (3,1) & (1, –1) satisfies but (3, –1) does not

76. [1] P : for all $M > 0$, there exists $x \in S$ such that $x \geq M$.
~ P : there exits $M > 0$, for all $x \in s$
Such that $x < m$
Negation of 'there 'exsits' is 'for all'.

77. [2]

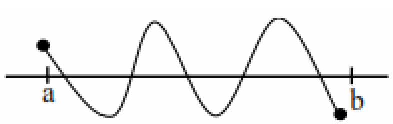


$r = \sqrt{6^2 + (3\sqrt{5})^2}$
$= \sqrt{36 + 45} = 9$

78. [2] $\vec{a} = (\vec{b}.\vec{c})\,\vec{b} - (\vec{b}.\vec{b})\vec{c}$
$= 1.2\cos\theta\,\vec{b} - \vec{c}$
$\Rightarrow \vec{a} = 2\cos\theta\,\vec{b} - \vec{c}$
$|\vec{a}|^2 = (2\cos\theta)^2 + 2^2 - 2.2\cos\theta\,\vec{b}.\vec{c}$
$\Rightarrow 2 = 4\cos^2\theta + 4 - 4\cos\theta.\,2\cos\theta$
$\Rightarrow -2 = 4\cos^2\theta$
$\Rightarrow \cos^2\theta = \frac{1}{2}$
$\Rightarrow \sec^2\theta = 2$
$\Rightarrow \tan^2\theta = 1$
$\Rightarrow \theta = \frac{\pi}{4}$
$1 + \tan\theta = 2$

79. [4] $C = A^2 - B^2$; $|C| \neq 0$
$A^5 = B^5$ and $A^3B^2 = A^2B^2$
Now, $A^5 - A^3B^2 = B^5 - A^2B^3$
$\Rightarrow A^3(A^2 - B^2) + B^3(A^2 - B^2) = 0$
$\Rightarrow (A^3 + B^2)(A^2 - B^2) = 0$
Post multiplying inverse of $A^2 - B^2$:
$A^3 + B^3 = 0$

80. [3]



$f(x) = \int_a^x g(t)dt$
$f(x) \rightarrow 5$
$f'(x) \rightarrow 4$
$g(x) \rightarrow 4$
$g'(x) \rightarrow 3$

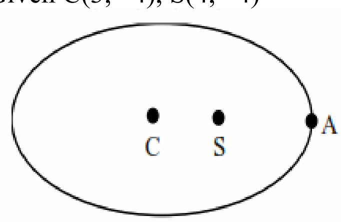
Section -B

81. [9] $\vec{a} = (1, -\alpha, \beta)$
$\vec{b} = (3, \beta, -\alpha)$
$\vec{c} = (-\alpha, -2, 1)$; $\alpha, \beta \in I$
$\vec{a}\vec{b} = -1 \Rightarrow 3 - \alpha\beta - \alpha\beta = -1$
$\Rightarrow \alpha\beta = 2$
1 2
2 1
–1 –2
–2 –1
$\vec{b}.\vec{c} = 10$
$\Rightarrow -3\alpha - 2\beta - \alpha = 10$
$\Rightarrow 2\alpha + \beta + 5 = 0$
$\therefore \alpha = -2;\ \beta = -1$
$$[\vec{a}\,\vec{b}\,\vec{c}] = \begin{vmatrix} 1 & 2 & -1 \\ 3 & -1 & 2 \\ 2 & -2 & 1 \end{vmatrix}$$
$= 1(-1 + 4) - 2(3 - 4) - 1(-6 + 2)$
$= 3 + 2 + 4 = 9$

82. [7] $\overrightarrow{QR} : -\ \frac{x-3}{1} = \frac{y+4}{-1} = \frac{z+5}{-6} = r$
$\Rightarrow (x,y,z) \equiv (r + 3, -r - 4, -6r - 5)$
Now, satisfying in in the given plane.
We get $r = -2$
so, required point of intersection is T(1, –2,7).
Hence, PT = 7.

83. [1] $\text{Re}(z) = \frac{3 - 6\cos^2\theta}{1 + 9\cos^2\theta} = 0.$
$\Rightarrow \theta = \frac{\pi}{4}$
Hence, $\sin^2 3\theta + \cos^2\theta = 1$

84. [3] Given C(3, –4), S(4, –4)



and A(5, –4)

Hence, a = 2 & ae = 1

$\Rightarrow e = \frac{1}{2}$

$\Rightarrow b^2 = 3$

So, E: $\frac{(x-3)^2}{4} + \frac{(y+4)^2}{3} = 1$

Intersecting with given tangent.

$\frac{x^2 - 6x + 9}{4} + \frac{m^2x^2}{3} = 1$

Now, D = 0 (as it is tangent)

So, $5m^2 = 3$.

85. [5] $I = 2\int_0^{\pi/2} \sin^3 x e^{-\sin^2 x}\, dx$

$= 2\int_0^{\pi/2} \sin x e^{-\sin^2 x} dx +$

$\int_0^{\pi/2} \underset{I}{\cos x}\ \underbrace{e^{-\sin^2 x}(-\sin 2x)}_{II} dx$

$= 2\int_0^{\pi/2} \sin x e^{-\sin^2 x}\, dx + \left[\cos x e^{-\sin^2 x}\right]_0^{\pi/2}$

$+ \int_0^{\pi/2} \sin x e^{-\sin^2 x} dx$

$= 3\int_0^{\pi/2} \sin x e^{-\sin^2 x} dx - 1$

$= \frac{3}{2}\int_{-1}^{0} \frac{e^{\alpha} d\alpha}{\sqrt{1+\alpha}} - 1 (\text{Put} - \sin^2 x = t)$

$= \frac{3}{2e}\int_0^1 \frac{e^x}{\sqrt{x}} dx - 1 (\text{Put } 1+\alpha = x)$

$= \frac{3}{2e}\int_0^1 e^x \underset{II}{\frac{1}{\sqrt{x}}} dx - 1$

$= 2 - \frac{3}{e}\int_0^1 e^x \sqrt{x} dx$

Hence, $\alpha + \beta = 5$

86. [2] $t^4 - t^3 - 4t^2 - t + 1 = 0$, $e^x = t > 0$

$\Rightarrow t^2 - t - 4 - \frac{1}{t} + \frac{1}{t^2} + 0$

$\Rightarrow \alpha^2 - \alpha - 6 = 0$, $\alpha = t + \frac{1}{t} \geq 0$

$\Rightarrow \alpha = 3, -2$ (reject)

$\Rightarrow t + \frac{1}{t} = 3$

$\Rightarrow$ The number of real roots = 2

87. [2] $\int e^{-y} dy = \int e^{\alpha x} dx$

$\Rightarrow e^{-y} = \frac{e^{\alpha x}}{\alpha} + c$ ....(i)

Put (x,y) = (ℓn2, ℓn2)

$\frac{-1}{2} = \frac{2^{\alpha}}{\alpha} + C$ ....(ii)

Put (x,y) ≡ (0,– ℓn2) in (i)

$-2 = \frac{1}{\alpha} + C$ ....(iii)

(ii) – (iii)

$\frac{2^{\alpha} - 1}{\alpha} = \frac{3}{a}$

$\Rightarrow \alpha = 2$ (as $\alpha \in N$)

88. [924] $N = 2^{10} \times 5^{10} \times 11^{11} \times 13^{13}$

Now, power of 2 must be zero,
power of 5 can be anything,
power of 13 can be anything
But, power of 11 should be even.
So, required number of divisors is
1 × 11 × 14 × 6 = 924

89. [832] B – C ≡ {7, 13, 19, ...97, ...}

Now, $n^2 - n \leq 100 \times 100$

$\Rightarrow n(n-1) \leq 100 \times 100$

⇒ A = {1,2, ..., 100}.

So, A ∩ (B – C) = {7, 13, 19, ...,97}

Hence, sum = $\frac{16}{2}$ (7 + 97) = 832

90. [2020] $A^n = \begin{vmatrix} 1 & n & \frac{n^2+n}{2} \\ 0 & 1 & n \\ 0 & 0 & 1 \end{vmatrix}$

So, required sum

$= 20 \times 3 + 2 \times \left(\frac{20 \times 21}{2}\right) + \sum_{r=1}^{20}\left(\frac{r^2 + r}{2}\right)$

= 60 + 420 + 105 + 35 × 41 = 2020